

HEAVY METALS STATUS IN VEGETABLES AND SOIL IRRIGATED WITH WASTEWATER IN DOMIZI CAMP, DUHOK, KURDISTAN REGION-IRAQ.

RAMADHAN OMER HUSSAIN*, DILSHAD ABDULRAHMAN MOHAMMED**

*Dept. of Horticulture, College of agricultural engineering sciences, University of Duhok, Kurdistan Region-Iraq.

**Directorate of the Environment of Duhok City, Kurdisatn Region-Iraq.

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ABSTRACT

This study was conducted in Domiz1 camp, Duhok, Kurdistan Region-Iraq to estimate the levels of heavy metals accumulation in soil and vegetables irrigated with contaminated water. The concentration of Pb, Cd, Co, and Zn measured in soil, wastewater, and fruits of tomato (*Solanum lycopersicum*), of eggplant (*Solanum melongena*), pepper (*Capsicum annum*), pumpkin (*Cucurbita moschata*), squash (*Cucurbita pepo*), gourd (*Lagenaria siceraria*), and okra (*Abelmoschus esculentus*). The results showed that the concentration of Pb, Cd, Co, and Zn in wastewater used for irrigation was (0.001, ND, 0.0015, 0.058 mg.l⁻¹) respectively; in soil was (6.8, 0.8, 9.0, and 80 mg.kg⁻¹) respectively; and in the studied vegetables ranged from (3.73-11.43 mg.kg⁻¹) for Pb, (0.22-0.59 mg.kg⁻¹) for Cd, (0.03-0.22 mg.kg⁻¹) for CO, and (38.83-65.70 mg.kg⁻¹) for Zn. Bio concentration factor for Pb was (0.549-1.681), for Cd it was (0.271-0.733), for Co it was (0.271-0.733), and for Zn it was (0.485-0.821). The maximum daily intake of Pb, Cd, Co, and Zn from eating the studied vegetables ranged from (0.381-1.633), (0.023-0.073), (0.003-0.028), and (5.728-9.381) mg.kg⁻¹ body weight respectively.

KEYWORDS: Pb, Cd, Co, Zn, wastewater. Bio concentration factor and Maximum daily intake.

INTRODUCTION

Vegetables are among the most important food stuffs for peoples of mostly all countries, because they are a source of very important nutritional substance such as minerals, vitamins, fibers, and antioxidant. When vegetables grown in a polluted environment, it may accumulate different toxic materials among which are heavy metals.

Municipal waste water is being used for irrigation purposes in several countries around the world due to many reasons including; large waste water availability in these countries, problems encountering the disposal, and saving fresh water for other purposes (Avci, 2013). As a result, the demands for using wastewater for irrigation purposes have been increased dramatically in recent years (Song *et al.*, 2006). However application of waste water for irrigation leads to improve soil properties because of the organic matter and nutritional elements it contain (Ali *et al.*, 2012). But, on the other hand, using

wastewater for irrigation lead to accumulate toxic elements in the soil such as heavy metals (Kumar and Chandra, 2004).

Heavy metals are those elements that have atomic number greater than twenty and a specific gravity more than one (Lasat, 2000). They present in the environment in a very low concentration (ppm). Heavy metals are very dangerous; because they are not bioagridable, tend to accumulate in organism tissues, and magnification through food chains. Some heavy metals such as Cu and Zn are essential elements for plant growth, while others such as (Pb, and Cd) are harmful to plants and didn't incorporate in any biological process of plant cell (Parsons, 2003).

Heavy metals in soil occurs naturally from the parent materials driving and due to human activity (Bowen, 1966 & Fytianos *et al.*, 2001). Natural sources include (geological sources, volcanic eruptions, and aerosol of forest fires). The major anthropogenic sources of heavy metals are the burning of fossil fuels, application of

pesticides and fertilizers, power stations, transporting, smelting, waste incineration, and recycling operations (Intawongse and Dean, 2006).

Domiz 1 camp is one of the several camps in Kurdistan Region-Iraq, which was built to accommodate the Syrian refugees. The camp located in the Domiz area that belong to Semel district, Duhok city, Kurdistan Region-Iraq. The camp was established on April, 2012. There are 6397 families in the camp which contains 28698 individuals of the two genders (BRHA, 2019). The majority of these families are with agricultural expertise, some of them grow some vegetables using the wastewater of the camp for the irrigation. Therefore this study was conduct to determine the validity of this water for irrigation in terms of heavy metals accumulation in all of soil, water, and plant.

MATERIALS AND METHODS

Study site

The present study was conducted in summer of the year 2018 on some vegetables fields in Domiz1 camp. Domiz1 camp elevation is (406 m) (Latitude: 36°46 '59.4 "N) and (Longitude: 42°52' 52.9"E). The weather is dry and hot in summer and cold and rainy in winter. The range of the rainfall is 535 mm and temperature degrees range from (-5.5) to (48) °C (Directory of Meteorology and Seismology, 2019)

Different vegetables shown in Table (1) were selected from Domiz1 camp farms and analyzed for lead (Pb), cadmium (Cd), zinc (Zn) and cobalt (Co) concentration in edible parts. These vegetables are commonly consumed by people there, and very highly relevant to human nutrition and health.

Table (1): Local, English and scientific name of Studied Vegetables

No.	Local name	English name	Scientific name
1	Bajansork	Tomato	<i>Solanum lycopersicum</i>
2	Bajanreshk	Eggplant	<i>Solanem melongena</i>
3	Fifi	Pepper	<i>Capsicum annum</i>
4	Qulnde Zer	Pumpkin	<i>Cucurbita moschata</i>
5	Qulnd Kosa	Squash	<i>Cucurbita pepo</i>
6	Qulnde Silehi	Gourd	<i>Lagenaria siceraria</i>
7	Bamia	Okra	<i>Abelmoschus esculentus</i>

Collection and digestion of water samples

In present study two set of water samples were taken, each with three replicates at 5 minutes interval. One sample was collected for determination of physical and chemical water properties-using polyethylene bottles and kept cold until reaching the lab, and the second one used for heavy metals determination, which stored in polyethylene bottles (pre-washed with a solution of nitric acid and water (1:1) and acidified by adding some drops of nitric acid to a pH below (2) in order to minimize the precipitation and adsorption of heavy metals on container walls.

Samples digested by adding 5 ml of concentrated nitric acid (HNO₃) to 100 ml of wastewater samples (APHA, 1958), and boiled to a possible smallest volume but not dryness, boiling of water samples continued until the water

samples reached colorless stage. The digested samples diluted by distilled deionized water in a volumetric flask of 25ml to the mark. The diluted samples were filtered in a polyethylene bottles and stored at 4°C until the analysis to be carried out.

Collection and digestion of soil samples

Three samples from top soil (0-30 cm) were collected from the farms irrigated by waste water. Air dried then crushed, sieved through 2-mm sieve and stored in polyethylene bags in order to be digested afterward.

Sieved soil samples were grounded finely by a handle mill. 0.5 g of grounded soil was weighted and digested in a 50-ml conical flask by adding 10 ml of a mixture of sulfuric acid, nitric acid, and perchloric acid with a ratio of 3:1:1 respectively on a hot plate (200° C) in the digestion chamber (APHA, 1958 Heating continued until a clear

color obtained, then diluted with distilled deionized water in a 50-ml volumetric flask and filtered by the filter paper. Filtered solution stored in a polyethylene bottles and kept in the refrigerator until analysis.

Collection and digestion of vegetables fruits

Three replicates of the vegetables fruits were harvested from each plants when reached to maturity (field maturity), about one kg fresh fruit for each vegetable from different plants (which represent one replicate).

Collected fruits packed in a polyethylene bags and brought to the lab, rinsed very well by tap water then splashed gently with distilled water. All fruits were cut by a stainless knife to possible smallest volume pieces and air dried for two days followed by oven drying at 70°C for 48 hrs. After that, dried samples were grounded by a handle mill in order to be ready for digestion.

Vegetable samples were digested according to the procedure used by (Tandon, 1999). 0.5 g of finely grounded powder was wet digested in a 100-ml conical flask by adding 10 ml mixture of nitric acid and perchloric acid in a ratio of 9:4 on a hot plate in the digestion chamber (fume hood). Heating and digestion continued until the liquid become colorless. The liquid further heated to the volume 2-3 ml, then lifted beside to lose the heat, then diluted by distilled deionized water in a 50

ml volumetric flask. Finally, the diluted sample filtered by the filter paper and stored in a polyethylene bottles for measuring heavy metals.

Measurement of heavy metals in water, soil and fruits

Concentration of lead, cadmium, zinc, and copper measured in the samples of water, soil, and fruits were measured by atomic absorption spectrophotometer type (SHIMADUZ 7000) in the labs of Directorate of the Environment of Duhok City, Kurdistan Region-Iraq.

Bio concentration factor (BCF)

Biological accumulation factor was measured by the equation used by (Qu *et al.*, 2011).

$$BCF = \frac{Pb \text{ in plant}}{Pb \text{ in soil}}$$

Where:

Pb in plant is a Pb concentration ($\mu\text{g.g}^{-1}$ dry weight) in plant tissues (shoots, roots, or fruits)

Pb in soil is a Pb content in the studied soil ($\mu\text{g.g}^{-1}$)

Daily intake

Daily intake ($\mu\text{g.kg}^{-1}$) body weight of each heavy metal from eating recommended 0.40 kg of fresh vegetables in a day (Najagi *et al.*, 2017) was calculated in respect to moisture content in fruits of studied vegetables (Table 2). The mean human body weight is considering 70 kg.

Table (2): Percent of dry weight in 1 kg fresh fruits of studied vegetables.

Vegetables	Moisture content (%)*	Percent of dry weight in 1 kg fresh weight (%)
Tomato	95	5
Eggplant	92	8
Pepper	94	6
Pumpkin	90	10
Squash	95	5
Gourd	92	8
Okra	93	7

*Calculated during the process of vegetables drying.

Calculation of Daily Metal Intake

The daily metal intake for each heavy metal from eating tested vegetables was calculated according to the following equation:

$$\left\{ \frac{\text{Metal con.} \left(\frac{\text{mg}}{\text{kg}} \text{ D. wt} \right) * \text{Percent of dry weight in 0.4 kg fresh vegetable (table 2)}}{\text{body weight (70 kg)}} \right\} * 1000 \mu\text{g}$$

Statistical Analysis

The randomized completely block design (R.C.B.D.) was depended for statistical analyses (Al-Rawi and Khalaf-Alla, 2000), the Micro soft (SAS 2002) was used for data analysis. Analysis of variance (ANOVA) and the differences between various treatments means at 5% level were tested with Duncan Multiple Range test (Duncan, 1955).

RESULTS AND DISCUSSION

Physical and chemical characteristic of irrigation water

Data showed in table (3) represent characteristic of wastewater used for irrigation of the studied vegetables. The chemical parameters measured in wastewater were pH, electrical conductivity (EC), total dissolved solids (TDS), salinity, dissolved oxygen (DO), chemical oxygen demand (COD), biochemical oxygen demand (BOD), total nitrogen (TN), total phosphorous (TP), Pb, Cd, Co, and Zn.

As it was expected pH of wastewater (7.82) was within the tolerance limit (6.5-8.5) of pH in irrigation water according to the standards recommended by (FAO, 1992). The value of EC was (703 $\mu\text{s.cm}^{-1}$), which didn't exceed

Daily Metal Intake (DMI) ($\mu\text{g.kg}^{-1}$ body weight)=

permissible limit of (3000 $\mu\text{s.cm}^{-1}$), low EC of water means low salinity (Rattan *et al.*, 2005).

Also DO (0.86 mg.l^{-1}) and COD (108.6 mg.l^{-1}) were very low when compared with standard limits of (9 mg.l^{-1}) for DO and (500 mg.l^{-1}) for COD. While BOD (217.2 mg.l^{-1}) was more than twice the safe limit (100 mg.l^{-1}) of BOD in irrigation water (FAO, 1992).

With regards to total nitrogen and total phosphorous content which are considered as a primary macronutrients for plant growth (El-Beshbeshy and Sheris, 1998), the results showed considerable amount of these two elements in waste water. It was found that the concentration of total N and P in irrigation water were lower than the prescribed limits (30 and 20 mg.l^{-1}) of N and P in respectively according to standards of (FAO 1985).

Concerning the heavy metals status in wastewater used for irrigation of studied vegetables, the results showed that the concentration of Pb, Cd, Co, and Zn in irrigation water were (0.001, ND, 0.0015, and 0.058 mg.l^{-1}) respectively.

Depending on the standards prescribed by (FAO, 1985 and FAO, 1992) for irrigation, it is obvious that concentration of these metals was lower than standard limits.

Table (3): Physical and chemical properties of wastewater used for the irrigation.

Parameter	Unit	Value	Standards *
PH	---	7.82	6.5-8.5
EC	$\mu\text{s.cm}^{-1}$	703	<3000
TDS	mg.l^{-1}	490	2000
DO	mg.l^{-1}	0.86	<9
COD	mg.l^{-1}	108.6	80-500
BOD	mg.l^{-1}	217.2	100
T.P.	mg.l^{-1}	17.933	30
T.N.	mg.l^{-1}	26	20

Pb	mg.l ⁻¹	0.001	2
Cd	mg.l ⁻¹	ND	0.01
Co	mg.l ⁻¹	0.0015	0.05
Zn	mg.l ⁻¹	0.058	2

* Limits for irrigation water (FAO 1985 and 1992)

Physical and chemical characteristic of studied soil

Concentration of heavy metals in agriculture soil (mg.kg⁻¹ soil) in the studied area is represented in Table (4). Concentration of Pb, Cd, Co, and Zn in soil was 6.8, 0.8, 9.0, and 80.0 (mg.kg⁻¹ soil) respectively. The extent of Pb, Cd,

and Zn was within the safe limits of (25-500), (3-6), and (300-600) mg.kg⁻¹ soil respectively according (WHO/FAO, 2007) standards. Co concentration in soil was also found to be within the safe range (1-40 mg.kg⁻¹ soil) (Alloway, 1990). Accordingly this soil is suitable for growing and production of vegetables.

Table (4): Physical and chemical properties of the soil.

No.	Soil Parameter	Value	Unit
1	Electrical Conductivity (EC)	0.427	ds.m ⁻¹
2	Sodium adsorption ratio	0.48	%
3	Organic Material (O.M)	9.20	g.kg ⁻¹
4	Soil Texture Class	Clay	
5	Sand	70.98	g.kg ⁻¹
6	Silt	300.64	g.kg ⁻¹
7	Clay	610.40	g.kg ⁻¹
8	pH	8.4	-
9	Total nitrogen (N)	45.20	mg.kg ⁻¹
10	Total phosphorus (P)	5.10	mg.kg ⁻¹
11	Total potassium (K)	4.88	mg.kg ⁻¹
12	Calcium Carbonate (CaCO ₃)	20.40	%
13	Cat ion exchange capacity(CEC)	32.88	Cmol.100 g ⁻¹ soil
14	Total lead	6.8	mg.kg ⁻¹
15	Total cadmium	0.8	mg.kg ⁻¹
16	Total cobalt	9.0	mg.kg ⁻¹
17	Total zinc	80	mg.kg ⁻¹

Heavy metals concentration in fruits of studied vegetables

Heavy metals concentration in fruits of studied vegetables irrigated with wastewater is shown in Table (5). Lead concentration fluctuated irregularly in fruits of tested vegetables, it ranged from (3.73 mg.kg⁻¹) to (11.43 mg.kg⁻¹) in squash and okra fruits respectively, that was differed significantly from all other treatments.

Lead known for its bad effects on human health-Exposure to Pb affects negatively on infants, reproductive system, liver, kidney, and stomach (Järp, 2003). Lead concentration in fruits of tomato, eggplant, and pepper which belong to family of Solanaceae ranged from (4.10-4.40

mg.kg⁻¹), which was more than Pb concentration (0.168-0.532 mg.kg⁻¹) in same vegetables (Mosleh and Almagabig, 2013). However these results were in accordance with the results of (Demirezen and Aksoy 2006) who found (5.3-9.7 mg.kg⁻¹) of Pb in same vegetables.

Lead concentration in fruits of Cucurbitaceae family vegetables; pumpkin, squash, and gourd ranged from (3.73-3.99 mg.kg⁻¹). These results were more than the results of (Antonious *et al.*, 2010 and Zhou *et al.*, 2016), but much lower than the results of (Ozores-Hampton *et al.*, 1997 and Demirezen and Aksoy, 2006). The content of Pb in okra fruits that belong to family Malvaceae accumulated more significantly Pb than all other vegetables. The present results of Pb in okra fruits was slightly more than results of (Demirezen and

Aksoy, 2006) who recorded (10.70 mg.kg⁻¹) in fruits of okra, in other hand both of Abbass, *et al.*, (2010) and Youssef and Eissa, (2015) found (0.064 and 0.24 mg.kg⁻¹) of Pb in fruits of okra irrigated with sewage water respectively. Results of Pb concentration in fruits of all tested vegetables were more than the safe limits of Pb in vegetables (0.50-2.00 mg.kg⁻¹) stated by world health organization and food and agriculture organization (WHO/FAO, 1976). Accumulation of Pb in fruits of vegetables might come from irrigation water, from soil, and/or from air pollution (Gorbounov *et al.*, 2003 and Al-Enezi *et al.*, 2004).

Like lead, cadmium is also a very toxic element to human and animals, it cause blood and heart diseases (WHO, 1987). The Cd concentration was (0.28, 0.33, and 0.59 mg.kg⁻¹) in eggplant, pepper, and tomato respectively. These values were more than results of (Goboasa *et al.*, 2010; and Bigdeli and Seilsepour, 2008), but noticeably smaller than (16.61 and 40.20 mg.kg⁻¹) of Cd in eggplant recorded by (Guo *et al.*, 2013 and Topcuoğlu and Önal, 2018) respectively; Perveen *et al.*, (2012) recorded (8.4-61.4) and (9.70 mg.kg⁻¹) of Cd in tomato and pepper respectively.

Fruits of pumpkin, squash, and gourd accumulated respectively (0.22, 0.22, and 0.38 mg.kg⁻¹) of Cd. The present results of Cd were in line with the those of Ozores-Hampton *et al.*, (1997), when they recorded (0.29 mg.kg⁻¹) of Cd in squash fruits. Correspondingly, very high Cd was accumulated by fruits of squash (41.60 mg.kg⁻¹) (Perveen *et al.*, 2012), and also in contrast to our data, slightly high Cd content (0.60 mg.kg⁻¹) was noticed by Demirezen and Aksoy, (2006) in pumpkin fruits.

Fruits of okra accumulated (0.51 mg.kg⁻¹) of Cd in the present study, which was higher than results of (Opaluwa, *et al.* 2012) and (Youssef and Eissa, 2015) who obtained (0.07 and 0.27) mg.kg⁻¹ of Cd respectively in fruits of okra, but it was in close to the (0.58 mg.kg⁻¹), results of (Demirezen and Aksoy, 2006). Cadmium introduced to the soil, water, and plants by human activities. However, no Cd concentration detected in irrigation water analyzed in present study but, still it presents in the soil about 0.8 mg.kg⁻¹ soil (Table 4), because it is more available than other heavy metals since it is more soluble in the soil (Järp,

2003). Also, Cd pollutes soils as impurities from phosphate fertilizers, in some studies about 7 ppm of Cd detected in phosphorous fertilizers (Wild, 1994).

Plant growth not need only macronutrients in order to be healthy but also require essential micronutrients such as Fe, Cu, Zn, Co, Ni, Mn, Mo, Bo to complete its life cycle (Marschner, 2012). Cobalt naturally found throughout the environment and long-term exposure to Co cause asthma, fibrosis, liver and kidney congestion, and effect on heart (ATSDR, 1992).

The cobalt concentration in studied vegetables ranged from the (0.22 mg.kg⁻¹) in squash fruits to (0.03 mg.kg⁻¹) in pepper fruits. Few studies were conducted on the Co intake by plants. However, Bigdeli and Seilsepour, (2008) investigated Co contamination in some vegetables irrigated with wastewater, and found that the Co concentration in eggplant fruits was zero but the concentration of Co was (0.01 mg.kg⁻¹) in tomato fruits. In another study conducted by Opaluwa, *et al.*, (2012) found that the concentration of Co in okra fruits was (0.02-0.28 mg.kg⁻¹).

Zinc concentration, which is essential metal for plant growth was (56.13, 39.33, and 53.83 mg.kg⁻¹) in tomato, eggplant and pepper respectively. These results are in consistent with the results of Topcuoğlu and Önal, (2018) as they found (29.0-228.0 mg.kg⁻¹) of Cd in eggplant, and very similar to the results of Youssef and Eissa, (2015) when they detected (35-55 mg.kg⁻¹) of Zn in fruits of tomato. On other hand a lower Zn accumulation was recorded by Gogoasa *et al.*, (2010); (5.0 mg.kg⁻¹) in tomato, (5.2 mg.kg⁻¹) in eggplant, and (6.4 mg.kg⁻¹) in pepper.

The lower Zn held by pumpkin fruits, that stand on (38.83 mg.kg⁻¹) (lower Zn content), while squash held (62.67 mg.kg⁻¹) of Zn, although gourd accumulated (50.0 mg.kg⁻¹) Zn in the fruits, and okra fruits contained the uppermost Zn (65.70 mg.kg⁻¹). These results were higher as compared to those of Zhou *et al.*, (2016) that show (2.88 mg.kg⁻¹) Zn in Pumpkin, and also to those of Mosleh and Almagabig, (2013) in squash (0.36 mg.kg⁻¹). Regarding the Zn concentration in squash and okra, these results were in line with those of Youssef and Eissa, (2015), who found (40-60 mg.kg⁻¹) and (35-60 mg.kg⁻¹) Zn concentration in squash and okra, respectively.

Table (5): Heavy metals concentration (mg.kg⁻¹d.wt) in studied vegetables irrigated with wastewater in Domizl camp.

Vegetables	Heavy metals			
	Lead	Cadmium	Cobalt	Zinc
Tomato	9.40 ab	0.59 a	0.06 bc	56.13 ab
Eggplant	4.10 c	0.28 cd	0.15 ab	39.33 c
Pepper	9.13 b	0.33 bc	0.03 c	53.00 bc
Pumpkin	3.87 c	0.22 d	0.043 c	38.83 c
Squash	3.73 c	0.22 d	0.22 a	62.67 ab
Gourd	3.99 c	0.38 b	0.17 a	50.00 bc
Okra	11.43 a	0.51 a	0.06 bc	65.70 a
Mean	6.52	0.36	0.11	52.24
Safe limits	0.50-2.00*	0.02-1.00*	0.05-0.50#	60.00*

* WHO/FAO 1976

Misra and Mani, 1991

Bio concentration factor

Bio concentration factor (BCF) is a parameter illustrates the ability of plant to transfer the metals from the soil to plant. In (Table 6) BCF ranged from (0.549-1.681) for Pb, (0.271-0.733) for Cd, (0.003-0.19) for Co, and (0.485-0.821) for Zn.

The BCF value of Pb (1.681) was more than results of (Yadav *et al.*, 2013). In current study, the minimum value of BCF was found for Co (0.003) in fruits of pepper. The values of BCF for measured heavy metals in studied vegetables were in order of Pb > Zn > Cd > Co.

Table (6): Bio concentration factor of heavy metals in studied vegetables irrigated with wastewater in Domizl camp.

Vegetables	Bio concentration factor			
	Lead	Cadmium	Cobalt	Zinc
Tomato	1.382 ab	0.733 a	0.007 bc	0.702 ab
Eggplant	0.603 c	0.350 cd	0.017 ab	0.492 c
Pepper	1.343 b	0.413 bc	0.003 c	0.663 ab
Pumpkin	0.569 c	0.271 d	0.005 c	0.485 c
Squash	0.549 c	0.283 d	0.025 a	0.783 ab
Gourd	0.587 c	0.475 b	0.019 a	0.625 bc
Okra	1.681 a	0.638 a	0.007 bc	0.821 a
Mean	0.959	0.452	0.012	0.653

Maximum daily intake

In present study the maximum daily intake (MDI) of Pb, Cd, Co, and Zn from each studied vegetables were calculated from eating recommended daily intake of vegetables (400g/day) (Najagi *et al.*, 2017).

Table (7) shows that the MDI range of heavy metals in tested vegetables was as follow; for Pb it was (0.381-1.633 $\mu\text{g.kg}^{-1}$ body weight), for Cd it was (0.023-0.073 $\mu\text{g.kg}^{-1}$ body weight), for Co it was (0.003-0.028 $\mu\text{g.kg}^{-1}$ body weight), whereas concerning Zn, it was (5.728 -

9.381 $\mu\text{g.kg}^{-1}$ body weight). The values of MDI of studied heavy metals were in order of Zn > Pb > Cd > Co. The fruits of vegetables grown in Domiz 1 camp that irrigated with wastewater had a DMI of Pb, Cd, Co, and Zn below the provisional tolerable daily intake (4.00 and 1.00 $\mu\text{g.kg}^{-1}$ body weight) for Pb and Cd respectively (WHO, 1987). (250.00 $\mu\text{g.kg}^{-1}$ body weight) for Co (Najagi *et al.*, 2017); and (143.00 $\mu\text{g.kg}^{-1}$ body weight) for Zn (WHO, 1982).

Depending on the daily intake values of Pb, Cd, Co, and Zn, and in comparison with

(Provisional tolerable daily intake) PTDI of four metals a person of 70 kg body weight could safely consume 400 g of studied vegetables per day (Table6).

Table (7): Daily intake ($\mu\text{g.kg}^{-1}$ body weight) from eating one kg of fresh vegetables.

Vegetables	Daily intake (Based on an adult of 70 kg)			
	Lead	Cadmium	Cobalt	Zinc
Tomato	0.959 bc	0.060 b	0.007 b	5.728 c
Eggplant	0.669 cd	0.046 c	0.025 a	6.422 bc
Pepper	1.118 b	0.040 c	0.003 b	6.490 bc
Pumpkin	0.789 c	0.044 c	0.009 b	7.925 ab
Squash	0.381 d	0.023 c	0.023 a	6.395 bc
Gourd	0.651 cd	0.062 b	0.028 a	8.163ab
Okra	1.633 a	0.073 a	0.009 b	9.381 a
mean	0.886	0.050	0.015	7.215
Provisional tolerable daily intake	4.0 *	1.00 *	250.00 **	143.00 ***

*(WHO, 1987), **(Najagi *et al.*, 2017), ***(WHO, 1982)

CONCLUSION

Depending on the results of the study it can be concluded that the levels of Pb, Cd, Co, and Zn in soil and irrigation water was within the safe limits, therefore the soil is suitable for growing vegetables, but the irrigation water must be subjected to some treatments such primary treatment to reduce the values of EC, BOD, and total nitrogen. However, the daily intake values of all studied metals were within the safe limits, but the concentration of Pb exceeded the maximum permissible levels in vegetables.

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